

Brønsted acid (Brönsted acid)

A *molecular entity* capable of donating a *hydron* (proton) to a base, (i.e. a 'hydron donor') or the corresponding *chemical species*. For example: H₂O, H₃O⁺, CH₃CO₂H, H₂SO₄, HSO₄⁻, HCl, CH₃OH, NH₃.

See also *conjugate acid-base pair*.

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