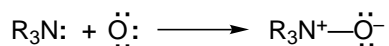


dipolar bond

A *bond* formed (actually or conceptually) by *coordination* of two neutral moieties, the combination of which results in charge-separated structures, e.g.



The term is preferred to the obsolescent synonyms ‘coordinate link’, ‘coordinate covalence’, ‘dative bond’, ‘semipolar bond’.

1994, 66, 1106