

**dissociation energy,  $E_d$ ,  $D$**

Energy required to dissociate a molecule into two parts. Subscripts 0 and e, *i.e.*  $D_0$  and  $D_e$ , are used to denote the dissociation from the ground state and potential energy minimum, respectively.

See also *bond-dissociation energy*.

G.B. 20