

**ionic strength,  $I$** 

On a molality basis,  $I_m = \frac{1}{2} \sum m_B z_B^2$  where the sum goes over all the ions B.  $z_B$  is the charge number of ion B. The ionic strength on a concentration basis is defined analogously  $I_c = \frac{1}{2} \sum c_B z_B^2$ .

G.B. 50; 1996, 68, 977